NADAR SARSWATHI COLLEGE OF ENGINEERING AND TECHNOLOGY, THENI.				
Course/Branch : BE / Civil	Year / Semester :IV / VII	Format No.	NAC/TLP-07a.13	
Subject Code :OML751	Subject Name : Testing of Materials	Rev. No.	02	
Unit No : IV	Unit Name : MATERIAL CHARACTERIZATION TESTING	Date	30.09.2020	

OBJECTIVE TYPE QUESTION BANK

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S. No.	Objective Questions (MCQ /True or False / Fill up with Choices)	BT L	
	Which of the following microscope are not used for microscopy?		
	a) Optical		
1	b) Ultrasonic	L1	
	c) Electron		
	d) Scanning probe		
	The resolving power of TEM is derived from		
2	a) electrons	ТЭ	
2	b) specifiens	LZ	
	d) ocular system		
	Which of the following component of TEM focuses the beam of electrons on the sample?		
	a) ocular lens		
3	b) condenser lens	L2	
5	c) stage	112	
	d) column		
	Image formation in electron microscope is based on		
	a) column length		
4	b) electron number	L1	
	c) differential scattering		
	d) specimen size		
	In SEM, the image is formed by the electrons that		
1.14	a) reflect back		
5	b) ionize	L2	
	c) undergo inversion		
	d) pass through		
	Which of the following is used in electron microscope?		
	a) electron beams	_	
6	b) magnetic fields	L1	
	c) light waves		
	d) electron beams and magnetic fields		
	Which among the following helps us in getting a three-dimensional picture of the specimen?		
-	a) Transmission Electron Microscope		
/	b) Scanning Electron Microscope	L2	
	d) Simple Microscope		
	Where do we obtain the magnified image of the specimen in SEM?		
	a) esthede ray tube		
8	h) phosphorescent screen	L2	
0	c) anode		
	d) scanning generator		
	X-ray diffractometers provide information about the compounds present in a solid sample.		
9	a) Quantitative		
	b) Qualitative	L3	
	c) Quantitative and qualitative		
	d) Either quantitative or qualitative		
Prep	ared By:R SANTHASEELAN, AP/Mech Page 1 o	f 3	

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10	Etching of specimen is c a) Visible grain bound b) Invisible grain bound c) Toughness d) Hardness	one to achieve ury ury		L2	2
11	 Which of the following aromatic conjugation wi a) Fourier transform infi b) Differential scanning c) U/V Visible spectros d) Thermo gravimetric a 	echnique is used to measure the number of conj hin the various molecules? ared calorimetry copy nalysis	jugated double bonds	and L1	1
12	 Fourier transform infrare a) Quantitative determin b) Curing and degrada c) Determination of vola d) Analysis of structural 	d spectroscopy is used to study ation of additives in polymers tion behaviour of cross-linked polymers tilities of plasticizers imperfections on the surface		L2	2
13	Atomic force microscop a) Spectral analysis b) Thermal analysis c) Mechanical testing d) Morphological analysis	y comes under the category of	3	L1	1
14	If the absorption of elect or longer wavelengths for a) Luminescence b) Fluorescence c) Phosphorescence d) Spontaneous emission	romagnetic radiation by matter results in the em r a short time, the phenomenon is termed as wh	nission of radiation of nich of the following?	the same	3
15	 a) Luminescence b) Fluorescence c) Phosphorescence d) Spontaneous emission 	y by an atom ionised by a higher energy X-ray	is a type of which of	the L2	2
16	The measurement of inte a) Destructive, quantitat b) Non-destructive, qua c) Destructive, qualitativ d) Non-destructive, qual	nsity of fluorescent X-rays provide a simple and ve intitative e itative	d way of	_ analysis.	3
17	The problem of spectrala) Trueb) False	interference is not severe in X-ray spectroscopy	7.	L2	2
18	In X-ray fluorescence sp fluorescence does not de a) Spectrum of the incid b) Angle of radiance c) Molecular weight d) Incident angle	ectrometer, the relationship between the excitat pend on which of the following? ent radiation	ion intensity and the	intensity of L2	2

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19	Fluorescent X-ray spectron a) True b) False	neters would require only moderate-intensity X-r	ay tubes.	L1
20	Sample recovery is possibl a) True b) False	e after spectroscopic analysis because the sample	is not chemically	v affected.
21	Bragg's law is not a suffici a) True b) False	ent condition for diffraction by crystalline solids.		L2
22	The transition zone for Rar a) Between vibrational ar b) Between electronic leve c) Between magnetic level d) Between magnetic level	nan spectra is d rotational levels s s of nuclei s of unpaired electrons.		L2
23	 Which of the following is a a) Structural investigation b) Basis of understanding of c) Study of energetically ex d) All of the mentioned 	n application of molecular spectroscopy? of colors acited reaction products		L2
24	The region of electromagne a) Microwave b) Radio frequency c) Infrared d) UV-rays	etic spectrum for nuclear magnetic resonance is _		L2
25	X-ray diffraction patterns a a) They have very high energy b) They are electromagnetic c) Their wavelengths are d) Their high frequency energy	re used for studying crystal structure of solids be rgy, hence they can penetrate through solids c radiation, and hence do not interact with matter comparable to inter-atomic distances ables rapid analysis	cause (crystals)	L2
26	The secondary electrons ra a) specimen b) anode c) vacuum chamber d) cathode	diated back in scanning microscope is collected b	py?	L1
27	Electron Microscope can g a) 400,000X b) 100,000X c) 15000X d) 100X	ive a magnification up to		L1